Control of Microbial Growth through Physico-Chemical Parameters

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Control of Microbial Growth: Introduction

- Early civilizations practiced salting, smoking, pickling, drying, and exposure of food and clothing to sunlight to control microbial growth.
- Use of spices in cooking was to mask taste of spoiled food. Some spices prevented spoilage.
- In mid 1800s Semmelweiss and Lister helped developed <u>aseptic techniques</u> to prevent contamination of surgical wounds. Before then:
 - Nosocomial infections caused death in 10% of surgeries.
 - Up to 25% mothers delivering in hospitals died due to infection

- **Sterilization:** Killing or removing *all forms of microbial life* (including <u>endospores</u>) in a material or an object.
- Heating is the most commonly used method of sterilization.
- **Commercial Sterilization**: Heat treatment that kills endospores of *Clostridium botulinum* the causative agent of botulism, in canned food.
- Does not kill endospores of thermophiles, which are not pathogens and may grow at temperatures above 45°C.

- **Disinfection:** Reducing the number of pathogenic microorganisms to the point where they no longer cause diseases. Usually involves the removal of <u>vegetative</u> or <u>non-endospore forming</u> pathogens.
- May use physical or chemical methods.
 - **Disinfectant**: Applied to inanimate objects.
 - Antiseptic: Applied to living tissue (antisepsis).
 - **Degerming**: Mechanical removal of most microbes in a limited area. Example: Alcohol swab on skin.
 - **Sanitization**: Use of chemical agent on food-handling equipment to meet public health standards and minimize chances of disease transmission. E.g: Hot soap & water.

Sepsis: Comes from Greek for decay or putrid. Indicates bacterial contamination.

Asepsis: Absence of significant contamination.

Aseptic techniques are used to prevent contamination of surgical instruments, medical personnel, and the patient during surgery.

Aseptic techniques are also used to prevent bacterial contamination in food industry.

Bacteriostatic Agent: An agent that *inhibits* the growth of bacteria, but does not necessarily kill them. Suffix stasis: To stop or steady.

Germicide: An agent that kills certain microorganisms.

Bactericide: An agent that kills bacteria. Most do not kill endospores.

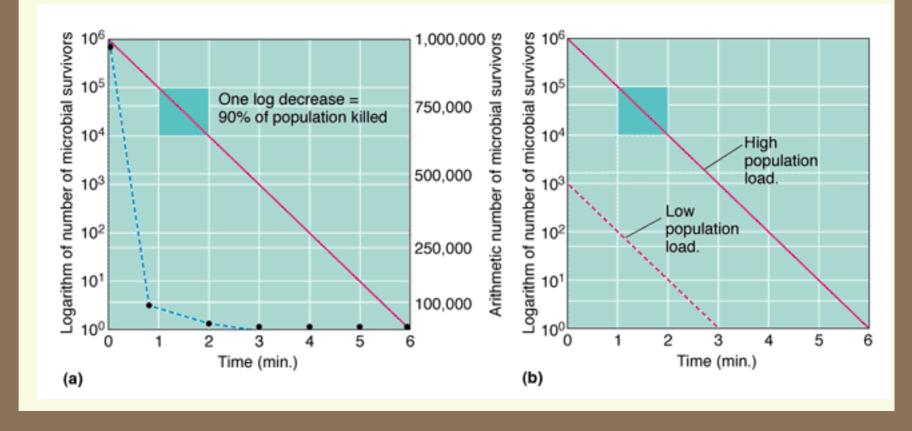
Viricide: An agent that inactivates viruses.

Fungicide: An agent that kills fungi.

Sporicide: An agent that kills bacterial endospores of fungal spores.

Control of Microbial Growth: Rate of Microbial Death

When bacterial populations are heated or treated antimicrobial chemicals, they usually die at a constant rate.



Control of Microbial Growth: Rate of Microbial Death

- Several factors influence the effectiveness of antimicrobial treatment.
 - **1. Number of Microbes**: The more microbes present, the more time it takes to eliminate population.
 - **2. Type of Microbes**: Endospores are very difficult to destroy. Vegetative pathogens vary widely in susceptibility to different methods of microbial control.
 - **3. Environmental influences:** Presence of organic material (blood, feces, saliva) tends to inhibit antimicrobials, pH etc.
 - **4. Time of Exposure:** Chemical antimicrobials and radiation treatments are more effective at longer times. In heat treatments, longer exposure compensates for lower temperatures.

- Heat: Kills microorganisms by denaturing their enzymes and other proteins. Heat resistance varies widely among microbes.
 - **Thermal Death Point (TDP)**: Lowest temperature at which all of the microbes in a liquid suspension will be killed in ten minutes.
 - **Thermal Death Time (TDT):** Minimal length of time in which all bacteria will be killed at a given temperature.
 - **Decimal Reduction Time (DRT):** Time in minutes at which 90% of bacteria at a given temperature will be killed. Used in canning industry.

- **Moist Heat:** Kills microorganisms by **coagulating** their proteins.
- In general, moist heat is much more effective than dry heat.
- **Boiling:** Heat to 100°C or more at sea level. Kills vegetative forms of bacterial pathogens, almost all viruses, and fungi and their spores within 10 minutes or less. Endospores and some viruses are not destroyed this quickly. However brief boiling will kill most pathogens.
 - Hepatitis virus: Can survive up to 30 minutes of boiling.Endospores: Can survive up to 20 hours or more of boiling.

Moist Heat (Continued):

Reliable sterilization with moist heat requires temperatures above that of boiling water.

Autoclave: Chamber which is filled with hot steam under pressure. Preferred method of sterilization, unless material is damaged by heat, moisture, or high pressure.

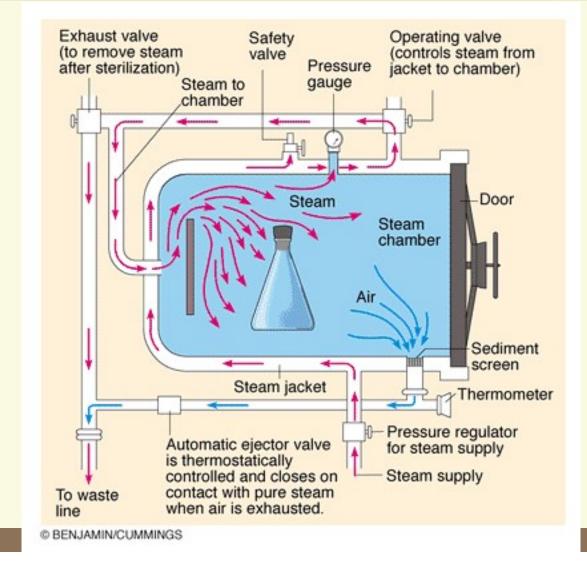
Temperature of steam reaches 121°C at twice atmospheric pressure.

Most effective when organisms contact steam directly or are contained in a small volume of liquid.

All organisms and endospores are killed within 15 minutes.

Require more time to reach center of solid or large volumes of liquid.

Autoclave: Closed Chamber with High Temperature and Pressure



Moist Heat (Continued):

Pasteurization: Developed by Louis Pasteur to prevent the spoilage of beverages. Used to reduce microbes responsible for spoilage of beer, milk, wine, juices, etc.

Classic Method of Pasteurization: Milk was exposed to 65°C for 30 minutes.

High Temperature Short Time Pasteurization (HTST): Used today. Milk is exposed to 72°C for 15 seconds.

Ultra High Temperature Pasteurization (UHT): Milk is treated at 140°C for 3 seconds and then cooled very quickly in a vacuum chamber.

Advantage: Milk can be stored at room temperature for several months.

Dry Heat: Kills by oxidation effects.

Direct Flaming: Used to sterilize inoculating loops and needles. Heat metal until it has a red glow.

Incineration: Effective way to sterilize disposable items (paper cups, dressings) and biological waste.

Hot Air Sterilization: Place objects in an oven.

Require 2 hours at 170°C for sterilization. Dry heat is transfers heat less effectively to a cool body, than moist heat.

- **Filtration:** Removal of microbes by passage of a liquid or gas through a screen like material with small pores. Used to sterilize heat sensitive materials like vaccines, enzymes, antibiotics, and some culture media.
 - **High Efficiency Particulate Air Filters (HEPA):** Used in operating rooms and burn units to remove bacteria from air.
- **Membrane Filters**: Uniform pore size. Used in industry and research. Different sizes:
 - **0.22 and 0.45um Pores**: Used to filter most bacteria. Don't retain spirochetes, mycoplasmas and viruses.
 - **0.01 um Pores**: Retain all viruses and some large proteins.

Low Temperature: Effect depends on microbe and treatment applied.

Refrigeration: Temperatures from 0 to 7°C. **Bacteriostatic effect**. Reduces metabolic rate of most microbes so they cannot reproduce or produce toxins.

Freezing: Temperatures below 0°C.

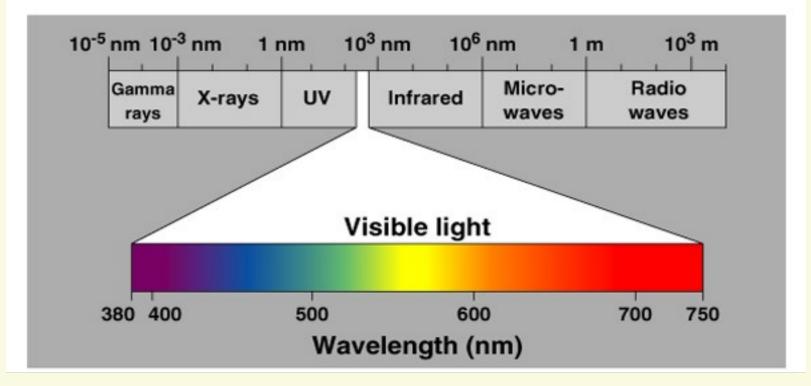
Flash Freezing: Does not kill most microbes.

Slow Freezing: More harmful because ice crystals disrupt cell structure.

Over a third of vegetative bacteria may survive 1 year. Most parasites are killed by a few days of freezing.

- **Radiation:** Three types of radiation kill microbes:
- **1. Ionizing Radiation:** Gamma rays, X rays, electron beams, or higher energy rays. Have short wavelengths (less than 1 nanometer).
- Dislodge electrons from atoms and form ions.
- Cause mutations in DNA and produce peroxides.
- Used to sterilize pharmaceuticals and disposable medical supplies. Food industry is interested in using ionizing radiation.
- *Disadvantages:* Penetrates human tissues. May cause genetic mutations in humans.

Forms of Radiation



Physical Methods of Microbial

Control:

Radiation: Three types of radiation kill microbes:

2. Ultraviolet light (Nonionizing Radiation): Wavelength is longer than 1 nanometer. Damages DNA by producing thymine dimers, which cause mutations.

Used to disinfect operating rooms, nurseries, cafeterias.

Disadvantages: Damages skin, eyes. Doesn't penetrate paper, glass, and cloth.

Chemical Methods of Microbial Control

Types of Disinfectants

1. Phenols and Phenolics:

Phenol (carbolic acid) was first used by Lister as a disinfectant.

Rarely used today because it is a skin irritant and has strong odor. Used in some throat sprays and lozenges.

Acts as local anesthetic.

Phenolics are chemical derivatives of phenol

Cresols: Derived from coal tar (Lysol).

Biphenols (pHisoHex): Effective against gram-positive staphylococci and streptococci. Used in nurseries. Excessive use in infants may cause neurological damage.

Destroy plasma membranes and denature proteins.

Advantages: Stable, persist for long times after applied, and remain active in the presence of organic compounds.

- **2. Halogens:** Effective alone or in compounds.
- A. Iodine:
 - **Tincture of iodine** (alcohol solution) was one of first antiseptics used.
 - Combines with amino acid tyrosine in proteins and denatures proteins.
 - Stains skin and clothes, somewhat irritating.
 - **Iodophors**: Compounds with iodine that are **slow releasing**, take several minutes to act. Used as skin antiseptic in surgery. Not effective against bacterial endospores.
 - Betadine
 - Isodine

2. Halogens: Effective alone or in compounds.**B. Chlorine:**

When mixed in water forms **hypochlorous** acid:

$$Cl_2 + H_2O ---->H+ + Cl- + HOCl$$

Hypochlorous acid

Used to disinfect drinking water, pools, and sewage.

Chlorine is easily inactivated by organic materials.

Sodium hypochlorite (NaOCI): Is active ingredient of bleach.

Chloramines: Consist of chlorine and ammonia. Less effective as germicides.

3. Alcohols:

Kill bacteria, fungi, but not endospores or naked viruses. Act by denaturing proteins and disrupting cell membranes. Evaporate, leaving no residue.

Used to mechanically wipe microbes off skin before injections or blood drawing.

Not good for open wounds, because cause proteins to coagulate.

Ethanol: Drinking alcohol. Optimum concentration is 70%.

Isopropanol: Rubbing alcohol. Better disinfectant than ethanol. Also cheaper and less volatile.

4. Heavy Metals:

Include copper, selenium, mercury, silver, and zinc.

Oligodynamic action: Very tiny amounts are effective.

A. Silver:

1% silver nitrate used to protect infants against gonorrheal eye infections until recently.

B. Mercury

Organic mercury compounds like merthiolate and mercurochrome are used to disinfect skin wounds.

C. Copper

Copper sulfate is used to kill algae in pools and fish tanks.

Chemical Methods of Control

Types of Disinfectants

4. Heavy Metals:

D. Selenium

Kills fungi and their spores. Used for fungal infections.

Also used in dandruff shampoos.

E. Zinc

Zinc chloride is used in mouthwashes.

5. Quaternary Ammonium Compounds (Quats):

Widely used surface active agents.

Cationic (positively charge) detergents.

Effective against gram positive bacteria, less effective against gram-negative bacteria.

Also destroy fungi, amoebas, and enveloped viruses.

Zephiran, Cepacol, also found in our lab spray bottles.

Pseudomonas strains that are resistant and can grow in presence of Quats are a big concern in hospitals.

Advantages: Strong antimicrobial action, colorless, odorless, tasteless, stable, and nontoxic.

Diasadvantages: Form foam. Organic matter interferes with effectiveness. Neutralized by soaps and anionic detergents.

6. Aldehydes:

Include some of the most effective antimicrobials. Inactivate proteins by forming covalent crosslinks with several functional groups.

A. Formaldehyde gas:

Excellent disinfectant.

Commonly used as *formalin*, a 37% aqueous solution.

Formalin was used extensively to preserve biological specimens and inactivate viruses and bacteria in vaccines.

Irritates mucous membranes, strong odor.

Also used in mortuaries for embalming.

Chemical Methods of Control Types of Disinfectants 6. Aldehydes:

B. Glutaraldehyde:

Less irritating and more effective than formaldehyde. One of the few chemical disinfectants that is a <u>sterilizing</u> <u>agent</u>.

A 2% solution of glutaraldehyde (Cidex) is:

Bactericidal, tuberculocidal, and viricidal in 10 minutes. Sporicidal in 3 to 10 hours.

Commonly used to disinfect hospital instruments.

Also used in mortuaries for embalming.

7. Gaseous Sterilizers:

Chemicals that sterilize in a chamber similar to an autoclave. Denature proteins, by replacing functional groups with alkyl groups.

A. Ethylene Oxide:

Kills all microbes and endospores, but requires exposure of 4 to 18 hours.

Toxic and explosive in pure form.

Highly penetrating.

Most hospitals have ethylene oxide chambers to sterilize mattresses and large equipment.

8. Peroxygens (Oxidizing Agents):

Oxidize cellular components of treated microbes. Disrupt membranes and proteins.

A. Ozone:

Used along with chlorine to disinfect water.

Helps neutralize unpleasant tastes and odors.

More effective killing agent than chlorine, but less stable and more expensive.

Highly reactive form of oxygen.

Made by exposing oxygen to electricity or UV light.

Chemical Methods of Control

Types of Disinfectants

8. Peroxygens (Oxidizing Agents):

B. Hydrogen Peroxide:

Used as an antiseptic.

Not good for open wounds because quickly broken down by catalase present in human cells.

Effective in disinfection of inanimate objects.

Sporicidal at higher temperatures.

Used by food industry and to disinfect contact lenses.

C. Benzoyl Peroxide:

Used in acne medications.

8. Peroxygens (Oxidizing Agents):

D. Peracetic Acid:

One of the most effective liquid sporicides available. **Sterilant** :

Kills bacteria and fungi in less than 5 minutes.

Kills endospores and viruses within 30 minutes.

Used widely in disinfection of food and medical instruments because it does not leave toxic residues.

Efficiency of Different Chemical Antimicrobial Agents

